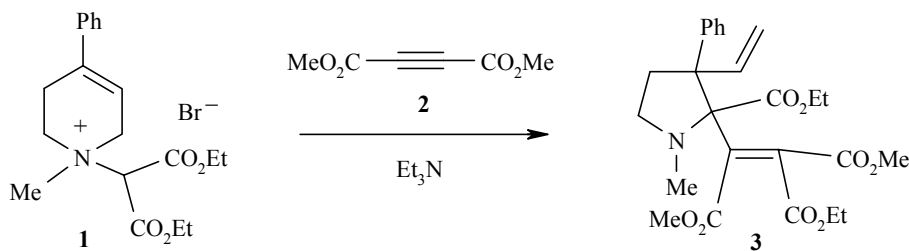


## UNEXPECTED COURSE FOR THE REACTION OF 1-DI(ETHOXYCARBONYLMETHYL)-1-METHYL- 4-PHENYL-1,2,3,6-TETRAHYDROPYRIDINIUM BROMIDE WITH DIMETHYL ACETYLENEDICARB- OXYLATE IN THE PRESENCE OF TRIETHYLAMINE

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Using acetylenedicarboxylic ester in the presence of base it has been possible to show for the first time that reaction of 2,3-dihydro-1H-indeno[2,1-*c*]pyridinium [1] and 1,2,3,4-tetrahydro- $\gamma$ -carbolinium [2] quaternary salts to tetrahydroindenoazonines and hexahydroazoninoindoles respectively can occur with expansion of the six membered  $\Delta^3$ -piperidine ring (with an alkoxy carbonylmethyl substituent on the nitrogen atom) to a nine membered ring [1, 2]. In this work we have studied the reaction of 1-di(ethoxycarbonylmethyl)-1-methyl-4-phenyl-1,2,3,6-tetrahydropyridinium bromide (1) with dimethyl acetylenedicarboxylate (2) in the presence of triethylamine under analogous conditions (20°C, 32 h). However, in place of the expected azonine derivative, column chromatography of the reaction mixture unexpectedly gave a 25% yield of the substituted pyrrolidine 3, the 2-ethenyl group of which contains three alkoxy carbonyl substituents. Its structure was



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unambiguously proved by X-ray crystallographic analysis, from its  $^1\text{H}$  and IR spectra, and by chromato-mass spectrometry (detailed data from the X-ray structural analysis will be presented in a separate report).

Such a surprising conversion of salt **1** to pyrrolidine **3** involves migration of an ethoxycarbonyl group and this can be rationalized by the following reaction steps. The electrophilic acetylene dicarboxylate ester **2** adds to the initially formed N-ylide and the 1,4-zwitterion formed evidently undergoes a 1 $\rightarrow$ 3 acyl shift of one of the ethoxycarbonyl groups to the carbanion center. This shift leads to a new ylide which can undergo a [3,2]-sigmatropic rearrangement with recyclization of the piperidine ring to a pyrrolidine.

The  $^1\text{H}$  NMR spectra were recorded on a Bruker WP-400 (400 MHz) spectrometer using DMSO- $d_6$  (compound **1**) or  $\text{CDCl}_3$  (compound **3**) with the residual deuterated solvent protons as internal standard. IR Spectra were taken on an Infracalum FT-801 spectrometer for KBr tablets. An Agilent 1100 liquid chromatograph with DAD, ELSD Sedex 75 detectors combined with an Agilent LC/MSD VL mass spectrometer with electrospray ionization was used to monitor the reaction mixtures and the purity of the separated compound **3**. X-ray structural analysis of compound **3** was carried out by a direct method on a Bruker SMART 1000 CCD diffractometer with  $\text{MoK}\alpha$  radiation, graphite monochromator, and with  $\theta$ - and  $\omega$ -scanning. The starting 1-methyl-4-phenyl-1,2,3,6-tetrahydropyridine was prepared as in method [6].

**1-Di(ethoxycarbonyl)methyl-1-methyl-4-phenyl-1,2,3,6-tetrahydropyridinium Bromide (1).** Bromo-malonate ester (2.8 g, 10 mmol) was added to a solution of 1-methyl-4-phenyl-1,2,3,6-tetrahydropyridine (2.1 g, 10 mmol) in absolute THF (20 ml) and the mixture was held with refluxing under a nitrogen atmosphere for 3 h. The cooled reaction mixture was treated with hexane (50 ml) and the precipitate formed was separated and recrystallized from acetone to give compound **1** with mp 78°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1753 and 1741 sh (C=O), 1629 (C=C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm ( $J$ , Hz): 1.27 (6H, t,  $J = 7.2$ ,  $\text{OCH}_2\text{CH}_3$ ); 2.92 (2H, m, H-3); 3.35 (3H, s,  $\text{NCH}_3$ ); 3.86 (2H, m, H-2); 4.33 (4H, m,  $\text{OCH}_2\text{CH}_3$ ); 4.41 and 4.53 (each 1H, both br. d,  $J = 16.5$ , H-6); 6.03 (1H, s,  $\text{NCHCOO}$ ); 6.15 (1H, br. s, H-5); 7.18-7.61 (5H, m,  $\text{C}_6\text{H}_5$ ). Found, %: Br 19.52; N 3.42.  $\text{C}_{19}\text{H}_{26}\text{BrNO}_4$ . Calculated, %: Br 19.42; N 3.39.

**2-(2-Ethoxycarbonyl-1-methyl-3-phenyl-3-vinylpyrrolidin-2-yl)-1-ethoxycarbonyl-1,2-di(methoxycarbonyl)ethene (3).** Triethylamine (1.8 ml, 14 mmol) was added to a suspension of the quaternary salt **1** (2.0 g, 5 mmol) and acetylenedicarboxylic ester **2** (1.5 g, 10 mmol) in absolute dioxane (30 ml) and stirred at 20°C under a nitrogen atmosphere for 32 h. Solvent was evaporated *in vacuo* and the residue was column chromatographed on  $\text{SiO}_2$  in the system hexane-ethyl acetate using a gradient from 1: 0 to 1: 10 to give compound **3** (0.67 g, 25%) as colorless crystals with mp 112-113°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1731 sh, 1727 (C=O), 1631 (C=C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm ( $J$ , Hz): 1.20 (3H, t,  $J = 7.6$ ,  $\text{OCH}_2\text{CH}_3$ ); 1.30 (3H, t,  $J = 7.6$ ,  $\text{OCH}_2\text{CH}_3$ ); 2.24 (3H, s,  $\text{NCH}_3$ ); 2.68 (2H, m, H-4); 2.86 and 3.33 (each 1H, both m, H-5); 3.45 (3H, s,  $\text{OCH}_3$ ); 3.52 (1H, s,  $\text{OCH}_3$ ); 4.08 and 4.22 (each 2H, both q,  $J = 7.6$ ,  $\text{OCH}_2\text{CH}_3$ ); 4.72 (1H, d,  $J = 17.2$ ,  $\text{CH}=\text{CH}$ -cis H-*trans*); 5.05 (1H, d,  $J = 10.8$ ,  $\text{CH}=\text{CH}$ -cis H-*trans*); 6.94 (1H, dd,  $J = 17.2$  and 10.8,  $\text{CH}=\text{CH}$ -cis H-*trans*); 7.17-7.58 (5H, m,  $\text{C}_6\text{H}_5$ ). Mass spectrum,  $m/z$ : 474  $[\text{M}+\text{H}]^+$ . Found, %: C 63.12; H 6.71; N 3.02.  $\text{C}_{25}\text{H}_{31}\text{NO}_8$ . Calculated, %: C 63.42; H 6.55; N 2.96. M 473.

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